

# Polymerizable channel-like stacks derived from cyclic tetrameric diacetylenes

Oleg V. Kulikov<sup>a,b,\*</sup>, Arshad Mehmood<sup>c</sup>, Yulia V. Sevryugina<sup>c</sup>

<sup>a</sup> Huntsman Advanced Materials, The Woodlands, TX, 77381, USA

<sup>b</sup> Department of Chemistry, Massachusetts Institute of Technology, Cambridge, MA, 02139, USA

<sup>c</sup> Department of Chemistry, Texas Christian University, Fort Worth, TX, 76129, USA

## ABSTRACT

A small series of cyclic diacetylenes has been synthesized and tested by scanning electron microscopy (SEM), single crystal X-ray diffraction, and atomic force microscopy (AFM) to examine their ability to form arrangements of discrete molecules in the solid state. Examining solid state structures clearly demonstrated propensity of diacetylene tetramers to form channel-like stacks. Remarkably, cyclotetradiyne **DA-4** monomer tends to produce two types of nano-tubular motifs (*i.e.*, the channel-like arrangement with solvent matrix inside and unprecedented flattened nano-tube incorporated no solvent guest molecules) which is indicative of the weak C–H/ $\pi$  interactions dominating and directing the stacking in such simplistic diacetylene cyclic systems. Noteworthy, **DA-4** molecules may be crystallized from organic solvents into birefringent branchy patterns revealed by Polarized Optical Microscopy (POM) and twisted ribbons of both handedness as evidenced by SEM data. We hypothesize that all these secondary structures may arise from the bundling the individual stacks discovered by single crystal X-ray analysis. Overall, engineering the elongated motifs may open up new opportunities for the future applications such as a rational design of synthetic ion channels and carriers, synthesis of the covalent polymeric nanotubes, *etc.*

## 1. Introduction

Ability of the various macrocycles and macroheterocycles to form channel-like motifs is well-established [1–12] which also includes stabilization of assemblies with functional polydiacetylenes. [13] Among those, an interesting and promising class of macrocycles prone to solid state stacking and topochemical polymerization is represented by dehydro [24]annulenes [11] and cyclic phenylacetylenes polymerizable under pressure [14]. Over the years, there has been an overwhelming interest in developing various polydiacetylenes (PDAs) based platforms [15], *i.e.*, gels, films, vesicles, mostly due to their unique optical properties [16] which allowed to use them in a large variety of applications, such as colorimetric sensors (blue/red phase switching) [17], colorimetric supramolecular gels [18a], diacetylene based multivalent glycol-polymers for high affinity lectin binding [18b], sensors as quality indicators in food packaging [19], chemosensors for colorimetric detection of cadmium ions [20], electromechanical actuators [21a], ultrastrong bioorthogonal Raman probes for targeted live-cell Raman imaging [21b], highly sensitive PDA ensembles for biosensing and bio-imaging [21c]. In addition to that, this widely investigated family of conjugated polymers attracted much attention due to their remarkable semiconducting and optoelectronic properties [22]. Interestingly,

topochemical polymerization of diacetylenes can be tuned and is closely related to the length of peripheral alkyl side groups exhibiting odd/even effect on polymerization [23]. In contrast to numerous examples of the linear PDA backbones described above, slightly less ubiquitous cyclic oligodiyne monomers may provide an access to polydiacetylene scaffolds bearing no peripheral side chains. Thus, understanding the underlying factors which impact cyclic diacetylenes' assembly in solid state, such as effect of inner and outer alkyl chains, crystallization solvent, appeared to be critical for successful topochemical polymerization. For instance, authors demonstrated the viability of synthetic approach towards graphene nanoribbons by using 1,4-diarylbutadiyne monomers that are initially converted to polydiacetylenes via topochemical polymerization in the crystal with subsequent cyclization and aromatization steps finally leading to graphene nanoribbons [24a]. Also, researchers [24b] described the solid cyclically-bound ladder polymers obtained from cyclic tetramers with no peripheral alkyl side chains. Thus, crystal-controlled 1,4-addition polymerization of cyclic tetramer afforded hydrocarbon organic nanotubes with carbon atoms connectivity favorable for the next extrusion step leading to carbon nanotube [24c].

In the present work, a series of cyclic diacetylenes (**DA-3–DA-6**) has been prepared and characterized by means of scanning electron microscopy (SEM) and single crystal X-ray analysis (X-ray) in a context of both

\* Corresponding author. Huntsman Advanced Materials, The Woodlands, TX, 77381, USA.

E-mail addresses: [Oleg.Kulikov@huntsman.com](mailto:Oleg.Kulikov@huntsman.com), [okulikov@mit.edu](mailto:okulikov@mit.edu) (O.V. Kulikov).

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UV- and thermally induced topochemical polymerization. Herein, we report synthesis and solid state polymerization of some cyclic oligo-diyne, namely, cyclotetracos-1,3,9,11,17,19-hexayne (**DA-3**), cyclodo triaconta-1,3,9,11,17,19,25,27-octayne (**DA-4**), cyclotetraconta-1,3,9,11,17,19,25,27,33,35-decayne (**DA-5**) which hold a great potential for constructing tunnel-like arrangements as a result of non-covalent CH  $\cdots$   $\pi$  interactions. Structurally similar diacetylene scaffolds were shown to be specifically aligned with respect to each other which is an important prerequisite for topochemical polymerization [25]. We reasoned that using aforementioned cyclic diacetylenes to construct polymerizable channel-like frameworks is advantageous as their molecules have no bulky substituents preventing tight packing of individual channels in a bundle, thus, giving a rise to highly ordered arrangements which is beneficial for high density PDA materials. In particular, **DA-4** cyclotetradecayne monomer [26a] represents considerable interest as its molecules are prone to self-organize into stackable motif in the solid state (e. g., needlelike crystals obtained from 25% (v/v) CHCl<sub>3</sub>-petroleum ether by room temperature evaporation) which was further polymerized<sup>26b,c</sup> using 50 Mrad of <sup>60</sup>Co  $\gamma$ -radiation. Likewise, abovementioned diacetylene tetramer, some alternative synthetically made backbones have shown to afford columnar polydiacetylenes having two parallel PDA chains that run parallel down opposite sides of a channel [27a] which may be of interest for the formation of highly conductive organic materials. This manuscript details morphological characterization of uncommon class of cyclic diacetylenes bearing no peripheral substituents (Scheme 1, Figure S1), thus, representing significant interest due to their ability to self-organize in a crystal lattice into tightly packed bundled channel-like arrangements, which consequently influences higher order organization, such as right and left-handed ribbons and fiber-like secondary PDA structures.

## 2. Results and discussion

### 2.1. Single crystal X-ray studies of channel-forming DA-4 macrocycle

To understand assembly factors leading to columnar arrangements, we investigated solid state structures of **DA-3**–**DA-5** cyclic diacetylenes. We presume that cyclic tetramer diacetylenes with C=C–C=C bond segments may be stacked as was shown for cyclic tetradecayne by growing crystals from specific solvent mixture [26b]. Having interstitial solvent molecules inside the stack of **DA-4** diacetylenes seemed to be important. We discovered in this study that solid state packing of **DA-4** macrocycles can be tuned by using different solvents. Thus, macrocyclic building blocks were shown to form extended, stackable arrays when crystallized from toluene, mesitylene, 1,4-dioxane, and CHCl<sub>3</sub>/EtOH solvent mixture (Fig. 1, S2–S22). For all structures inspected by X-ray, measured packing parameters are somewhat close to repeat distance  $d_1 = 4.9$  Å, the contact distance  $d_2 = 3.5$  Å, and tilt angle  $\theta = 45^\circ$  reported for butadiyne topochemical polymerization [11] as well as for single-crystal-to-single-crystal transformation of diacetylenes to PDA [27a],  $d_1 = 4.99$  Å,  $d_2 = 3.58$  Å, and tilt angle  $\theta = 41.6^\circ$ , correspondingly. Those parameters are also reminiscent of distances found in crystal structure of macrocyclic naphthalene diacetylene nanotubes [27b] ( $d_1 = 5.29$  Å,  $d_2 = 3.79$  Å, and tilt angle  $\theta = 45.65^\circ$ ). Remarkably, macrocyclic diacetylene-terthiophene plate-like polymerizable co-crystal [27c]

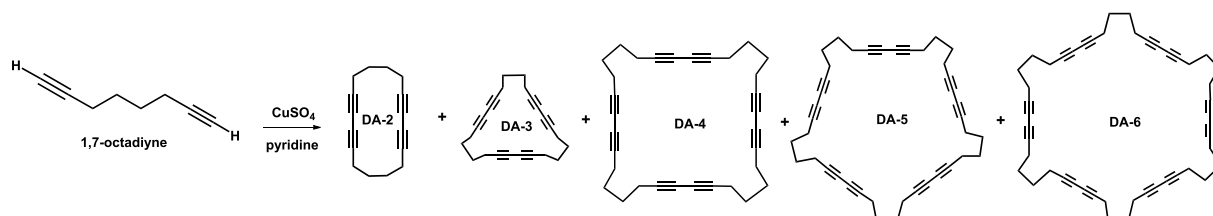
showed appreciably greater distance values, i.e.,  $d_1 = 10.31$  Å,  $d_2 = 8.12$  Å, and tilt angle  $\theta = 45.4^\circ$ , which are still considered to be close-to-ideal packing parameters [27b] preferred for topochemical polymerization. Adjusting the size of macrocyclic diacetylenes to ensure perfect columnar stacking is certainly a challenge in the crystal engineering. Thus, authors [27d] successfully constructed photo-irradiation polymerizable tubular assembly from L-glutamic acid and *trans*-1,4-cyclohexanediol macrocyclic diacetylene compound with no peripheral aromatic segments as the result of intermolecular hydrogen bonding.

Single crystal X-ray studies revealed that individual **DA-4** diacetylenes in the crystal lattice are pre-organized in a way favorable for topochemical polymerization with  $d_1$  ranging from 4.94 to 5.03 Å,  $d_2$  ranging from 3.42 to 3.69 Å and tilt angle  $\theta = 43.3$ – $47.8^\circ$ . We attribute unremarkable differences in diacetylene geometries to the crystal packing forces due to the solvent effect. All the **DA-4** structures demonstrated non-planar conformation (Figures S2, S5, S9, and S13). Stacking the discrete cyclic diacetylenes afforded channel-like structures with interstitial solvent molecules occupying the free volume inside those arrangements (Fig. 1, S3, S4, S6–S8, S10–S12).

We attribute this to weak C–H/ $\pi$  as well as van der Waals interactions which may be stabilizing channel-like arrangement. The latter may optionally incorporate template solvent molecules.

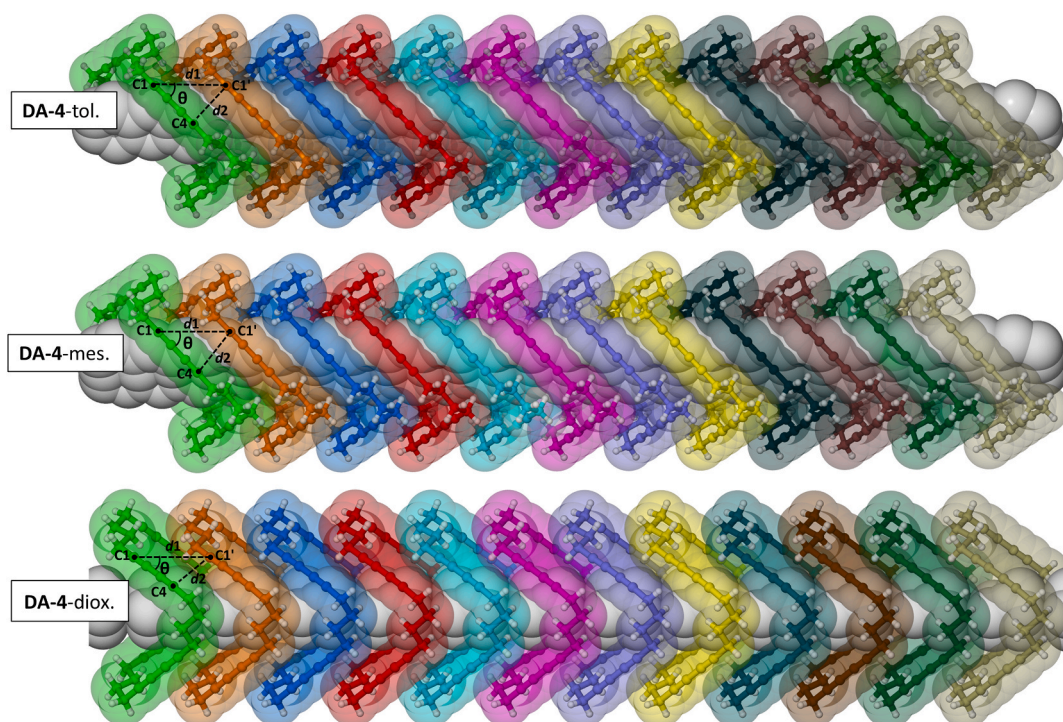
Apparently, **DA-4** channel-like motif obtained from CHCl<sub>3</sub>/EtOH solvent mixture is vastly different from those outlined in Fig. 1 as no template solvent molecules were detected inside the assembly (Fig. 2, panels a and b). Therefore, CHCl<sub>3</sub>/EtOH derived nanotube is somewhat flattened compared to the rest of **DA-4** assemblies obtained from other solvents. Since AFM specimen was prepared using the same EtOH/CHCl<sub>3</sub> stock solution from which crystals for single crystal X-ray analysis were grown, we infer that organization of **DA-4** molecules in the tubular morphologies at nano-scale level may mimic their assembly in crystal lattice. Furthermore, partial crystal packing diagram suggested that dozens of individual nanotubes can be bundled into fiber-like aggregates with diameter less than 25 nm easily identifiable on AFM phase diagram (Fig. 2, panels c, d; S23). More nano-tubular stacking diagrams can be found in ESI (Figures S4, S7, S8, S11, S12, S15), thus, providing clear evidence that such bundles is a common motif for **DA-4** type diacetylenes, at least, in the crystal lattice. Given the proximity of adjacent C<sub>1'</sub> and C<sub>4</sub> carbon atoms as well as geometrical orientation of the individual macrocycles favorable for formation of the new double bond (Figs. 1 and 2), polymerization seemed likely to occur under irradiation or at high temperatures.

Thus, the choice of crystallization solvent plays a significant role in constructing and stabilizing the **DA-4** monomer reactive phase. Analysis of a variety of self-assembled structures examined by single crystal X-ray analysis is suggestive that **DA-4** diacetylenes are prone to topochemical polymerization. Unlike aforementioned structures, packing diagrams of **DA-3** and **DA-5** diacetylenes did not reveal apparent channel-like constructs (Figures S17–S19, S21–S22) which proved that **DA-4** diacetylenes have optimal geometry and cycle dimensions that enable stacking of individual rings into the nano-tubes. We speculate that **DA-3** molecule represents constrained geometries whereas both **DA-5** and **DA-6** diacetylenes with much larger cavity have many more degrees of conformational freedom which complicates pre-alignment of the neighboring diacetylene units needed for columnar organization. For

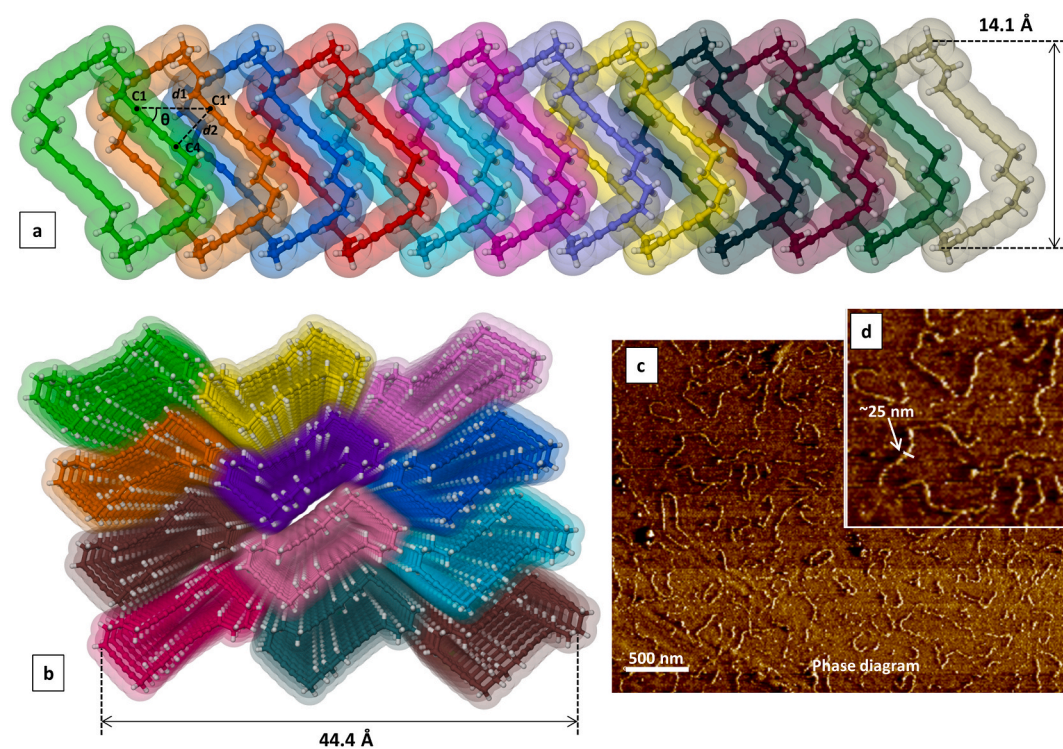


Scheme 1. Synthesis of cyclic diacetylenes from 1,7-octadiyne using oxidative coupling reaction.





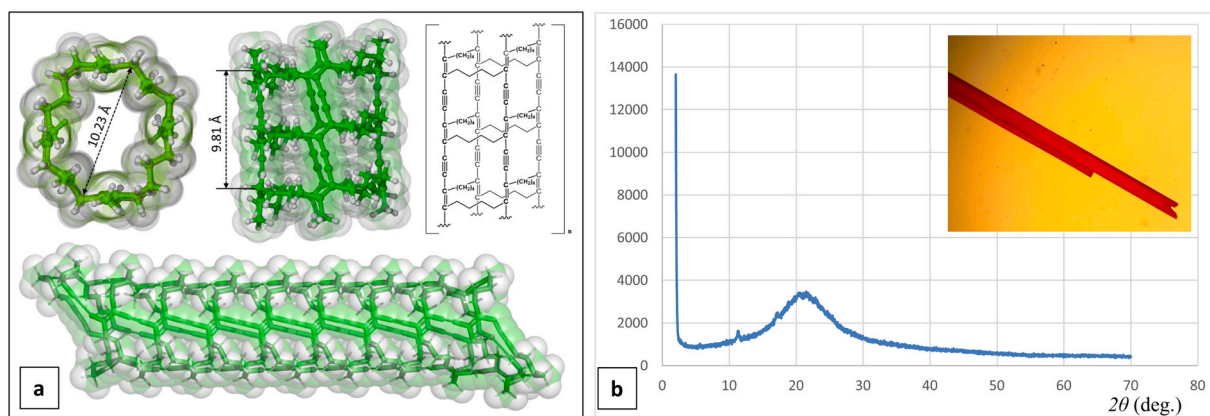
**Fig. 1.** Shown here are channel-like motifs assembled from twelve individual segments (cycles) as potentially polymerizable structures (disordered solvent molecules inside the assemblies shown as white overlapping spheres). Arrangement parameters, **DA-4** from toluene, CCDC 1852360:  $\theta = 44.6^\circ$ ,  $d_1 = 5.03 \text{ \AA}$ ,  $d_2 = 3.54 \text{ \AA}$ ; **DA-4** from mesitylene, CCDC 1852358:  $\theta = 47.8^\circ$ ,  $d_1 = 4.94 \text{ \AA}$ ,  $d_2 = 3.69 \text{ \AA}$ ; **DA-4** from 1,4-dioxane, CCDC 1852361:  $\theta = 43.3^\circ$ ,  $d_1 = 4.99 \text{ \AA}$ ,  $d_2 = 3.42 \text{ \AA}$



**Fig. 2.** Shown here is **DA-4** cyclic monomer from  $\text{CHCl}_3/\text{EtOH}$ , CCDC 1852357:  $\theta = 44.4^\circ$ ,  $d_1 = 5.11 \text{ \AA}$ ,  $d_2 = 3.58 \text{ \AA}$ . The tubular “solvent-free” motif is formed from twelve individual diacetylene tetramers (panel a) which are packed in a channel-like fashion (panel b). Panel c represents AFM phase diagram of fibrous network obtained by depositing **DA-4** stock in  $\text{CHCl}_3/\text{EtOH}$  on Si-wafer and annealed at  $100^\circ \text{C}$ .







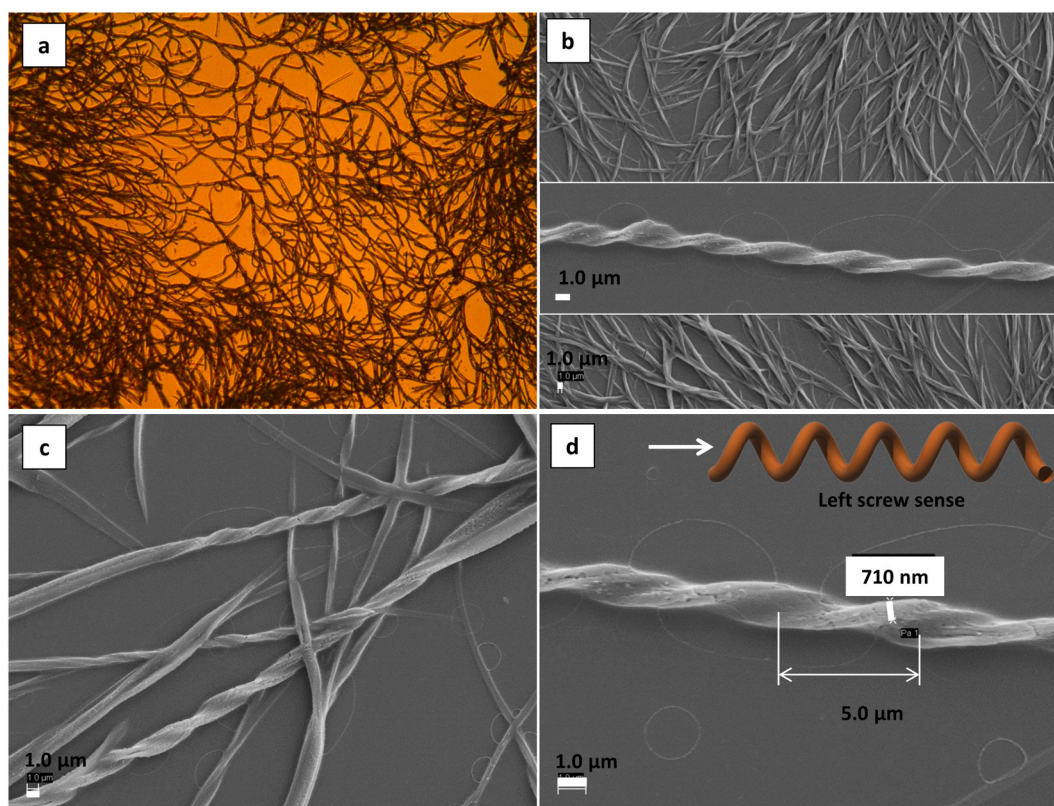
**Fig. 4.** Panel a: a hypothetical polydiacetylene architecture comprised of 3 individual DA-4 molecules, view atop and the side view (DFT geometry optimization, B3LYP, 3-21G basis set) and respective 9-cyclic stack (semi-empirical PM3 calculations) featuring alternating C=C double and C≡C triple bonds; panel b: pXRD plot of DA-4 (inset image represents UV-254 nm irradiated DA-4 red crystal).

techniques would be limited to single crystal X-ray analysis, Raman and IR-spectroscopies due to the very low solubility of polydiacetylene (PDA) material in common organic solvents [28]. Gratifyingly, pXRD measurements of UV-irradiated samples as well as their respective FT-IR spectra (broad C=C band at  $\sim 1700\text{ cm}^{-1}$ ) provided substantial evidence for polymerization of DA-4 monomer. Panel b of Fig. 4 exhibited one broadened first order reflection peak ( $2\theta \sim 20$  deg. with  $d$ -spacing  $\sim 4.4$  Å) which likely corresponds to the repeat distance between individual hydrocarbon cycles in PDA structure.

Calculations performed on polyDA-4 scaffold suggested the presence of cavity inside PDA framework as depicted above. We hypothesize that

void space may be filled with solvent guest molecules remaining after “crystal-to-crystal” polymerization. Such ladder-type polymer nanotubes with long conjugation chains may be useful to construct materials for sensing and electronics [25a].

Again, reactive DA-4 phases obtained by crystallization of diacetylene monomers from 1,4-dioxane and toluene seemed to be prone to 1,4-addition polymerization upon thermal annealing at elevated temperatures, e.g., 100, 140, and 210 °C, as evident by appearance of characteristic band at  $\sim 1700\text{ cm}^{-1}$  in their respective FT-IR spectra (Figures S32–S36).



**Fig. 5.** POM micrograph of partially polymerized DA-4 monomer deposited from 1,4-dioxane (panel a) and SEM images of DA-4 deposited from 1,2-dichloroethane (panels b–d, scale bar: 1.0 μm). Slides with deposited material were irradiated with UV-light (20 W, 365 nm, 2 h exposure) and washed with CHCl<sub>3</sub> to remove unreacted monomer.

### 2.3. PDA secondary structures investigated by POM, AFM, SEM, and TEM analytical techniques

Polarized optical microscopy (POM) images revealed densely packed fibrous branched patterns when DA-4 stock solutions were deposited on glass slides and allowed to evaporate slowly at ambient temperature (Fig. 5, panel a). We explored possibility of growing these specific intertwined crystals from different solvents including THF, 1,4-dioxane, and toluene (Figures S37–S39) and these morphologies were found to be predominant in a wide range of concentrations. Moreover, we examined aggregation behaviours of DA-4 diacetylene using SEM imaging (Fig. 5, panels b–d). Thus, we found out that polymerized DA-4 material from 1,2-dichloroethane may form distinctive secondary structures of both left and right handedness as shown below. More images of this nature are presented in ESI (Figures S40–S43).

It is uncertain at this point what caused the left- and right-handed helix screw sense in polymerized DA-4 diacetylene. We believe these interesting helical aggregates are both solvent and concentration dependent as changing the solvent from 1,2-dichloroethane to CHCl<sub>3</sub>/EtOH (10 wt%) binary mixture resulted in short bent fibers (Figure S44) which are somewhat reminiscent of PDA microcrystals obtained by spin-coating of diacetylene macrocycle on the Si wafer followed by thermal treatment as outlined previously [27a]. Other examples of PDA nanotubes and nanowires having bundled structure could be found in the literature [25c,25d]. Noteworthy, authors [27e] were able to acquire high resolution STM images of discrete PDA nanotubes from amphiphilic macrocyclic diacetylene with 3.52 nm width on the HOPG surface.

In fact, helical (or ribbon-like) DA-4 aggregates persist in a wide range of concentrations, i.e., from 1.25 mg/mL to 10 mg/mL. We hypothesize that higher concentrations caused more extensive aggregation and molecular ordering of DA-4 monomer units into helical assemblies. The helical pitch, or the distance between identical places on neighboring helical turns, can fluctuate, but estimated to be around ~5 μm with a thickness of helical turn of ~710 nm (Fig. 5d). It is not uncommon for acetylenic species to form chiral secondary structures. Thus, poly(phenylacetylene)s [29,30] are known to be helical as well as there are some precedents in the literature regarding chirality induction in polydiacetylenes [31]. Moreover, diacetylenes afforded helical nanoribbon polydiacetylenes via supramolecular gelation [32] and amphiphilic diacetylenes were found to self-assemble into vesicles and supramolecular gel with helical structure [33]. There is a reason to believe that DA-4 discrete molecules may similarly generate chiral ribbon-like structures of both handedness when deposited from 1,2-dichloroethane on a substrate surface which further underwent polymerization upon exposure to UV-light. Another plausible explanation of observed unusual aggregation behavior and molecular ordering involves “twisting upon crystallization” phenomenon when chirality imparted to achiral polymers by crystallization process [34]. The results of investigation carried out are suggestive that aggregation behavior of diacetylenic species can be tuned by changing the solvent and concentration of stock solution.

### 3. Conclusion

Thus, we experimentally confirmed by single crystal X-ray analysis that DA-4 tetrameric diacetylene obtained during a course of copper-mediated ring closing reaction is prone to self-organize into polymerizable channel-like architectures. Accordingly, we have obtained tubular covalent polydiacetylenes (PDAs), a family of conjugated polymers. Another important experimental finding was that DA-4 based nanotubes obtained from CHCl<sub>3</sub>/EtOH solvent mixture do not incorporate template solvent molecules unlike similar tubular assemblies obtained from toluene, mesitylene, and 1,4-dioxane. Finally, discovered polymerized DA-4 twisted ribbons as well as other secondary structures may result from specific arrangement of the individual DA-4 molecules followed by formation of ene-yne alternated PDA backbone upon

exposure to UV-light or high temperature. We assume that achiral DA-4 monomer may assemble into chiral secondary structures identifiable by SEM. Overall, control over fibrillar/ribbon aggregates thickness through varying DA-4 monomer concentration and solvent change may represent significant interest for potential technological applications as it provides unique, synthetic context to explore successful manipulation of PDA assemblies at molecular level, thus, leading to the synthesis of “carbon-rich” acetylenic materials.

### 4. Materials and methods

All starting materials were obtained from Sigma-Aldrich, Fluka or TCI America. Thin layer chromatography (TLC) was performed on Sigma-Aldrich TLC Plates (silica gel on aluminum, 200 mm layer thickness, 2–25 mm particle size, 60 Å pore size). Column chromatography was performed using silica gel (230–400 mesh) from Merck. <sup>1</sup>H nuclear magnetic resonance spectra were recorded at 500 MHz on Bruker Avance III™ 500 instrument at room temperature; <sup>1</sup>H chemical shifts are reported in parts per million (ppm) relative to the corresponding residual solvent peak. Mass-spectra were recorded using Accurate Mass QT of LC/MS instrument at University of Texas at Austin, Chemistry Department Mass Spectrometry Facility.

#### 4.1. X-ray single crystal structure determination of target molecules DA-3, DA-4 and DA-5

All data were collected using graphite-monochromated Mo Kα radiation (λ = 0.71073 Å). The structures were solved by direct methods and refined with SHELXL [35], Bruker APEX2 [36] and Olex2 [37,38] software packages: SHELXL2014/7 for DA-3, SHELXL2016/6 for DA-4 from 1,4-dioxane, SHELXL2014/7 for DA-4 from CHCl<sub>3</sub>/EtOH, SHELXL2014/7 for DA-4 from mesitylene, SHELXL2016/6 for DA-4 from toluene, SHELXL2014/7 for DA-5 from CHCl<sub>3</sub>. Crystallographic data have been deposited with the Cambridge Crystallographic Data Centre, CCDC 1852137 (DA-3), 1852138 (DA-5), 1852357 (DA-4 from CHCl<sub>3</sub>/EtOH), 1852358 (DA-4 from mesitylene), 1852360 (DA-4 from toluene), 1852361 (DA-4 from 1,4-dioxane). Copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (fax: +44-(0)1223-336033 or e-mail: deposit@ccdc.cam.ac.uk).

Single crystal X-ray of DA-3 from hexane/toluene. Structure: C<sub>24</sub>H<sub>24</sub>. Unit Cell Parameters: *a* = 22.2098(8), *b* = 8.4481(3), *c* = 10.1908(4), α = 90.00°, β = 98.697(2)°, γ = 90.00°, *V* = 1890.12(12) Å<sup>3</sup>, *M<sub>r</sub>* = 312.43, *Z* = 4, monoclinic, space group C 1 c 1. Mo Kα radiation (λ = 0.71073 Å), μ = 0.062 mm<sup>-1</sup>, *T* = 100(2)K. *R* (reflections) = 0.0348 (3885), *wR*<sup>2</sup>(reflections) = 0.0776(4339). The goodness of fit on *F*<sup>2</sup> was 1.031.

Single crystal X-ray of DA-4 from 1,4-dioxane (structure solution without a solvent). Structure: C<sub>16</sub>H<sub>16</sub> · 0.5(C<sub>32</sub>H<sub>32</sub>). Unit Cell Parameters: *a* = 24.023(3), *b* = 4.9890(6), *c* = 27.219(3), α = 90.00°, β = 112.475(3)°, γ = 90.00°, *V* = 3014.5(7) Å<sup>3</sup>, *M<sub>r</sub>* = 208.29, *Z* = 4, monoclinic, space group C 1 2 1. Mo Kα radiation (λ = 0.71073 Å), μ = 0.051 mm<sup>-1</sup>, *T* = 100(2)K. *R* (reflections) = 0.0791(5678), *wR*<sup>2</sup>(reflections) = 0.1993(6652). The goodness of fit on *F*<sup>2</sup> was 1.066.

Single crystal X-ray of DA-4 from CHCl<sub>3</sub>/EtOH. Structure: C<sub>32</sub>H<sub>32</sub>. Unit Cell Parameters: *a* = 5.1123(3), *b* = 8.1034(5), *c* = 15.6029(9), α = 101.341(2)°, β = 97.191(2)°, γ = 103.738(2)°, *V* = 605.52(6) Å<sup>3</sup>, *M<sub>r</sub>* = 416.57, *Z* = 1, triclinic, space group P -1. Mo Kα radiation (λ = 0.71073 Å), μ = 0.064 mm<sup>-1</sup>, *T* = 100(2)K. *R* (reflections) = 0.0633 (2305), *wR*<sup>2</sup>(reflections) = 0.1248(2708). The goodness of fit on *F*<sup>2</sup> was 1.173.

Single crystal X-ray of DA-4 from mesitylene (structure solution without a solvent). Structure: C<sub>32</sub>H<sub>32</sub>. Unit Cell Parameters: *a* = 4.9352(4), *b* = 26.4828(19), *c* = 11.2314(9), α = 90°, β = 91.449(3)°, γ = 90°, *V* = 1467.5(2) Å<sup>3</sup>, *M<sub>r</sub>* = 416.58, *Z* = 2, monoclinic, space group P 1 21/n 1. Mo Kα radiation (λ = 0.71073 Å), μ = 0.053 mm<sup>-1</sup>, *T* = 293(2)K. *R* (reflections) = 0.1262 (2019), *wR*<sup>2</sup>(reflections) = 0.2376(3191). The goodness of fit on *F*<sup>2</sup> was 1.069.



Single crystal X-ray of **DA-4** from toluene (structure solution without a solvent). Structure:  $C_{32}H_{32}$ . Unit Cell Parameters:  $a = 5.0299(5)$ ,  $b = 25.890(2)$ ,  $c = 11.0946(11)$ ,  $\alpha = 90^\circ$ ,  $\beta = 92.574(3)^\circ$ ,  $\gamma = 90^\circ$ ,  $V = 1443.3(2) \text{ \AA}^3$ ,  $M_r = 416.58$ ,  $Z = 2$ , monoclinic, space group  $P 1 21/n 1$ . Mo  $K\alpha$  radiation ( $\lambda = 0.71073 \text{ \AA}$ ),  $\mu = 0.054 \text{ mm}^{-1}$ ,  $T = 100(2) \text{ K}$ .  $R(\text{reflections}) = 0.0569(2517)$ ,  $wR^2(\text{reflections}) = 0.1400(3158)$ . The goodness of fit on  $F^2$  was 1.028.

Single crystal X-ray of **DA-5** from  $CHCl_3$ . Structure:  $C_{40}H_{40}$ . Unit Cell Parameters:  $a = 37.190(3)$ ,  $b = 8.3471(6)$ ,  $c = 10.0031(7)$ ,  $\alpha = 90^\circ$ ,  $\beta = 95.842(2)^\circ$ ,  $\gamma = 90^\circ$ ,  $V = 3089.1(4) \text{ \AA}^3$ ,  $M_r = 520.72$ ,  $Z = 4$ , monoclinic, space group  $C 1 c 1$ . Mo  $K\alpha$  radiation ( $\lambda = 0.71073 \text{ \AA}$ ),  $\mu = 0.063 \text{ mm}^{-1}$ ,  $T = 100.03 \text{ K}$ .  $R(\text{reflections}) = 0.0675(5929)$ ,  $wR^2(\text{reflections}) = 0.1603(6969)$ . The goodness of fit on  $F^2$  was 1.067.

#### 4.2. Synthesis and characterization of DA-3, DA-4, and DA-5 diacetylenes

Diacetylenes were prepared according to reported procedure using oxidative coupling of 1,7-octadiyne with  $CuSO_4$  in pyridine [26a,c].

To a 2 L three-neck round-bottomed flask equipped with a mechanical stirrer, thermometer, additional funnel, and reflux condenser, 225 g (1.14 mol) of cupric acetate monohydrate, 15.0 g (0.141 mol) of 1,7-octadiyne, and 1.5 L of pyridine were added. Reaction mixture allowed to stir at  $43^\circ \text{C}$  for 4 h, cooled, and filtered. The solids retained on the filter were washed with 100 mL of benzene (three times). The benzene washings were combined with the pyridine filtrate, and solvent was removed under reduced pressure to near dryness. The resulting dark brown residue was extracted with 1.5 L of 40% (v/v) water-benzene mixture. The benzene layer was separated from the aqueous layer and washed with 100 mL of water, 200 mL of 3 N HCl solution (2 times), 200 mL of water, and dried over  $MgSO_4$ . Concentration of the solution and drying under reduced pressure provided a dark brown solid which was subjected to a column of  $SiO_2$  using hexanes/benzene (4:1) mixture as an eluent. The desired cyclotetradiyne monomer (**DA-4**) was isolated by column chromatography along with **DA-2**, **DA-3**, **DA-5**, and **DA-6** cyclic diynes.

Dimeric **DA-2** (cyclohexadeca-1,3,9,11-tetrayne).  $^1H$  NMR ( $CDCl_3$ ):  $\delta$  2.21 (s, 8H), 1.73 (s, 8H); FT-IR (neat): 2948, 2931, 2900, 2859, 2253, 2186, 2140, 1714, 1441, 1454, 1422, 1362, 1318, 1273, 1236, 1176, 1035, 894, 795, 738, 576,  $459 \text{ cm}^{-1}$ . MS (CI)  $m/z$ : 208  $[M]^+$ .

Trimeric **DA-3** (cyclotetradeca-1,3,9,11,17,19-hexayne).  $^1H$  NMR ( $CDCl_3$ ):  $\delta$  2.27 (s, 12H), 1.67 (s, 12H); FT-IR (neat): 2929, 2897, 2862, 2828, 2255, 2164, 1454, 1422, 1366, 1322, 1273, 1029, 735,  $434 \text{ cm}^{-1}$ . MS (CI)  $m/z$ : 312  $[M]^+$ .

Tetrameric **DA-4** (cyclodotriaconta-1,3,9,11,17,19,25,27-octayne).  $^1H$  NMR ( $CDCl_3$ ):  $\delta$  2.26 (s, 16H), 1.68 (s, 16H); FT-IR (neat): 2947, 2921, 2898, 2860, 2826, 2255, 2165, 1756, 1454, 1422, 1366, 1321, 1255, 1170, 1032, 772, 726, 692, 626, 463,  $448 \text{ cm}^{-1}$ ; MS (high res. ESI)  $m/z$ : 524.16  $[M + Ag]^+$ .

Pentameric **DA-5** (cyclotetraconta-1,3,9,11,17,19,25,27,33,35-decayne).  $^1H$  NMR ( $CDCl_3$ ):  $\delta$  2.29 (s, 20H), 1.66 (s, 20H); FT-IR ( $CHCl_3$ ): 2939, 2866, 2260,  $2163 \text{ cm}^{-1}$ ; MS (CI)  $m/z$ : 521  $[M + 1]^+$ .

Hexameric **DA-6** (cyclooctatetraconta-1,3,9,11,17,19,25,27,33,35,41,43-dodecayne).  $^1H$  NMR ( $CDCl_3$ ):  $\delta$  2.29 (s, 24H), 1.64 (s, 24H); FT-IR ( $CHCl_3$ ): 2939, 2866, 2260,  $2163 \text{ cm}^{-1}$ . MS (CI)  $m/z$ : 626  $[M + 2]^{2+}$ .

#### 4.3. Attempted polymerization of DA-4 monomer using UV-254 handheld lamp and 365 nm (20 Watt) light source

Isolated single crystals (or material deposited on the glass slides and silicon wafers) were irradiated by UV-lamps (254 nm or 365 nm) for  $\sim 12$  h. Generally, the crystals were found unacceptable to run single crystal X-ray determination after long UV-light exposure. Solid material on glass slides (or silicon wafers) was gently rinsed with chloroform several times to remove unreacted monomer and soluble oligomers before preparing specimens for SEM analysis.

#### Credit author statement

Oleg V. Kulikov: Conceptualization, Methodology, Software, Formal analysis, Writing – original draft. Arshad Mehmood: Visualization, single crystal X-ray analysis (data collection) and structural refinement. Yulia V. Sevryugina: single crystal X-ray analysis and structural refinement.

#### Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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#### Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.rinma.2022.100262>.

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